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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
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EXAMINER

BARTON, JEFFREY THOMAS

ART UNIT

PAPER NUMBER

1795

MAIL DATE

DELIVERY MODE

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PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary	Application No. 10/721,929	Applicant(s) ANDERSON, CHRISTOPHER M.	
	Examiner Jeffrey T. Barton	Art Unit 1795	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 30 July 2008.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-40 is/are pending in the application.
- 4a) Of the above claim(s) 27-33 is/are withdrawn from consideration.
- 5) ☒ Claim(s) 7, 10 and 13-20 is/are allowed.
- 6) ☒ Claim(s) 1-6, 8, 9, 11, 12, 21-26, and 34-40 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|--------------------------------------------------------------------------------------|-------------------------------------------------------------------|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413) |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | Paper No(s)/Mail Date. _____ |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| Paper No(s)/Mail Date _____ | 6) <input type="checkbox"/> Other: _____ |

DETAILED ACTION

Comments

1. The amendment filed on 30 July 2008 lists "new" claims 27-33. This is improper, since different claims 27-33 were originally filed and withdrawn from consideration as directed to a nonelected invention following Applicant's election of 20 July 2007. In the interest of expediting prosecution, the Examiner will treat the application as follows:

- a. Original claims 27-33 (Submitted 24 November 2003) are considered to be pending and withdrawn from consideration, based on Applicant's election of 20 July 2007.
- b. The new claims, misnumbered 27-33 and presented in the amendment of 30 July 2008 are renumbered 34-40 and examined on the merits herein.

Status of Objections and Rejections Pending Since the

Office Action of 19 September 2007

2. The objection to claim 5 is withdrawn.
3. The rejections of claims 10, 11, and 13-20 under 35 U.S.C. §112, second paragraph are withdrawn due to Applicant's amendment.
4. All other rejections are maintained.

Claim Rejections - 35 USC § 112

5. The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

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6. Claims 9 and 40 rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

There is no positive antecedent basis for "said system controller" in line 3 of the claim. In addition, there is no positive antecedent basis for "said pH sensor" in line 4, "said water level sensor" in lines 3-4, "said water fill valve" in line 4, said "electrolyte fill valve" in line 4, and "said oxygen vent valve" in line 4. Antecedent basis for these five limitations would be provided by changing dependence to be on claim 7. Claim 40 depends from claim 9, and is therefore rejected on the same grounds.

Claim Rejections - 35 USC § 103

7. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

8. The factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

1. Determining the scope and contents of the prior art.
2. Ascertaining the differences between the prior art and the claims at issue.
3. Resolving the level of ordinary skill in the pertinent art.
4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

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9. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

10. Claims 1-6 and 21-26 are rejected under 35 U.S.C. 103(a) as being unpatentable over Cifaldi (US 6,372,978) in view of either Moroi et al (US 2002/0039534) or Araoka et al. (JP 09-139217)

Regarding claims 1, 5, and 21, Cifaldi discloses a system comprising a solar electrolysis power source comprising a solar panel (Figure 2, cells 46 on collector 10), an electrolysis unit including an electrolysis chamber/electrolyte tank containing aqueous electrolyte (Figure 3; Column 7, lines 19-47; electrolysis chamber/electrolyte tank 142); a hydrogen tank (Figure 3; tank 154); a hydrogen powered fuel cell (Column 7, lines 40-42); and a control unit including an inverter (Figure 2, inverter 138), a microprocessor (Figure 2; Within computer 96), and a modem (Figure 2, within computer 96; Column 6, lines 51-56); wherein said microprocessor is connected with said modem (Column 6, lines 51-56) and said inverter (Column 6, lines 46-56; Column 7, lines 52-58; Figure 2); and wherein said microprocessor controls said inverter. (Column 6, lines 46-56; Column 7, lines 52-58; Figure 2)

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Regarding claims 2, 5, and 21, Cifaldi discloses the inverter being connected to a power grid (Column 7, lines 52-58; Column 2, lines 43-45), which is typically monitored and controlled by a local power utility, and wherein the inverter is connected with an individual consumer (Column 7, lines 52-58; the building's occupant(s) correspond to such a consumer), and wherein said microprocessor is linked to said local power utility through said modem. (Column 6, lines 51-56; connection via phone line provides such a link)

Regarding claims 3 and 21, Cifaldi discloses the solar electrolysis power source producing and storing hydrogen gas, and providing hydrogen gas to the fuel cell. (Column 7, lines 19-31 and 40-44)

Regarding claim 6, Cifaldi discloses the electrolysis system including a source of water (conducted via 162); wherein said electrolysis unit is connected with said source of water (Figure 3) and receives water from said source of water, and wherein said electrolysis unit provides the electrolysis of said water and produces hydrogen gas and oxygen gas (Column 7, lines 19-46); wherein said solar panel is connected with said electrolysis unit and wherein said solar panel receives solar rays and provides electrical energy to said electrolysis unit (Column 7, lines 19-23); and wherein said hydrogen tank is connected to and received hydrogen gas from the electrolysis unit. (Column 7, lines 28-31)

Regarding claims 22-26, no particular weight can be given to the limitations in these claims, since they are entirely directed to intended use of the system defined in claim 21. Since Cifaldi's system is controlled by a computer connected to the system

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via a telephone line (Column 6, lines 46-60), it is perfectly capable of receiving control signals from an electric utility or any other desired entity having access to the telephone lines. Since the system is therefore capable of carrying out the intended function, the limitations do not render the claims patentable.

Cifaldi also teaches that automatic control of system functions from a central controller that is controlled by a computer is preferred, citing the ease of modification of the schemes. (Column 6, lines 46-60)

Cifaldi does not explicitly teach a hermetically sealed compressor, nor does he explicitly teach the microprocessor being connected to or controlling the hydrogen-powered fuel cell.

Moroi et al teach a hermetically sealed compressor used in compressing oxygen and hydrogen gases produced by hydrolysis for storage and later consumption in a fuel cell. (Abstract; Background; Summary sections; Paragraph 0037)

Araoka et al teach compressors 9a and 9b, used to compress oxygen and hydrogen gases produced by hydrolysis for storage and later consumption in a fuel cell. (English Abstract)

It would have been obvious to one having ordinary skill in the art at the time the invention was made to modify the system of Cifaldi by specifically connecting the computer 96 to the fuel cell disclosed as useful in the system, in order to control the fuel cell operation within the system, because Cifaldi teaches the advantage of automatic control of system components, in that modification of operation schemes is facilitated. (Column 6, lines 46-60)

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It would also have been obvious to one having ordinary skill in the art at the time the invention was made to incorporate a compressor between the hydrogen output of the electrolytic cell and the hydrogen storage tank, as taught by either Moroi et al or Araoka et al, because it would allow storage of the hydrogen gas at elevated pressures, reducing the tank size necessary for hydrogen storage. As evidenced by Moroi et al and Araoka et al, such compression means for the hydrogen produced in an electrolytic system is conventional, and the benefits of increased hydrogen storage density would have been clear to one having ordinary skill. In addition, hermetic sealing of such a compressor and any other component of a hydrogen gas processing system would have clearly been obvious, in order to prevent hydrogen leakage and the explosion hazard that would result.

Regarding the limitations "for containing water and a pH sensor" in claim 1 and "for containing water and a water level sensor" in claim 21, these are directed to the intended use of the device, and are therefore not given undue weight. A recitation of the intended use of the claimed invention must result in a structural difference between the claimed invention and the prior art in order to patentably distinguish the claimed invention from the prior art. If the prior art structure is capable of performing the intended use, then it meets the claim. See *In re Casey*, 152 USPQ 235 (CCPA 1967) and *In re Otto*, 136 USPQ 458, 459 (CCPA 1963). The Examiner considers the electrolysis chamber of Cifaldi to be capable of containing sensors as claimed, and the structure therefore meets the claim limitations.

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Regarding claim 4, selection of the output voltage and frequency of the inverter would depend on the requirements of the loads to be powered. Selection of any useful frequency and voltage would have been obvious to one having ordinary skill in the art, depending on this consideration.

11. Claim 8 is rejected under 35 U.S.C. 103(a) as being unpatentable over Cifaldi and either Moroi et al or Araoka et al as applied to claims 1-6 and 21-26 above, and further in view of Mérida-Donis.

Cifaldi in view of either Moroi et al or Araoka et al teaches systems as described above in addressing claims 1-6 and 21-26.

None among Cifaldi, Moroi et al, and Araoka et al explicitly teaches a hydrogen tank including a fill valve, output valve, and pressure gauge as claimed.

Mérida-Donis teaches a system for electrolytic hydrogen production and use in a fuel cell, comprising a storage means 706, which can be pressurized storage (Column 15, lines 56-58); a fill valve 820 between the hydrogen source and storage means 706; an output valve (regulator 854) located between the storage means and fuel cell 802; and a pressure gauge 818 indicating the pressure of the hydrogen gas in the storage means. (Column 15, lines 57-61)

It would have been obvious to one having ordinary skill in the art at the time the invention was made to further modify the system of Cifaldi and either Moroi et al or Araoka et al by providing the hydrogen tank with inlet and outlet valves, as well as a pressure gauge, as taught by Mérida-Donis, because Mérida-Donis teach the

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effectiveness of such means in controlling the flow, pressure, and consumption of hydrogen gas in such electrolytic hydrogen fuel cell systems. A skilled artisan would have recognized the value of such flow and pressure control means in a hydrogen gas system, in that they provide necessary control to maintain safe pressures of hydrogen in the system.

12. Claim 11 is rejected under 35 U.S.C. 103(a) as being unpatentable over Cifaldi and either Moroi et al or Araoka et al as applied to claims 1-6 and 21-26 above, and further in view of Caesar. (US 5,711,865)

Cifaldi in view of Moroi et al or Araoka et al is relied upon for the reasons given above.

None among Cifaldi, Moroi et al, or Araoka et al explicitly teaches a sulfuric acid electrolyte.

Caesar teaches an electrolytic hydrogen and oxygen gas-producing system in which sulfuric acid use used as the electrolyte. (Column 1, lines 48-52)

It would have been obvious to one having ordinary skill in the art at the time the invention was made to further modify the device of Cifaldi by specifically selecting sulfuric acid as the electrolyte, as taught by Caesar, because Caesar teaches that this is a preferable electrolyte for electrolytic hydrogen and oxygen production. Particularly since Cifaldi is silent concerning the particular electrolyte used in the system, a skilled artisan would have turned to the related prior art, such as Caesar, for selection of an appropriate material. Note also that the selection of a known material based on its

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suitability for its intended use supported a prima facie obviousness determination in *Sinclair & Carroll Co. v. Interchemical Corp.*, 325 U.S. 327, 65 USPQ 297 (1945).

13. Claim 12 is rejected under 35 U.S.C. 103(a) as being unpatentable over Cifaldi and either Moroi et al or Araoka et al as applied to claims 1-6 and 21-26 above, and further in view of the further teachings of Araoka et al.

Cifaldi in view of either Moroi et al or Araoka et al teaches systems as described above in addressing claims 1-6 and 21-26.

Neither Cifaldi nor Moroi et al explicitly teaches a water source that is a water tank holding water.

In addition to the teachings cited above, Araoka et al teach that water is provided to an electrolytic cell for hydrogen production from a water tank 3. (English Abstract)

It would have been obvious to one having ordinary skill in the art to modify the water line 162 of Cifaldi by specifically connecting it to a water tank, as taught by Araoka et al, because a water tank is known in the art to be a convenient means for holding water to be provided to an electrolytic cell as needed, as evidenced by Araoka et al. As water is consumed in the cell, it will necessarily need to be replenished, particularly in embodiments where the water generated by the fuel cell is not collected and returned to the electrolytic cell. A water tank is a conventional, convenient means for doing so. The results of such a modification would have been predictable and desirable.

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14. Claims 34-37 are rejected under 35 U.S.C. 103(a) as being unpatentable over Cifaldi and either Moroi et al or Araoka et al as applied to claims 1-6 and 21-26 above, and further in view of Kitajima et al. (US 5,833,831)

Cifaldi in view of either Moroi et al or Araoka et al is relied upon for the reasons given above.

None among Cifaldi, Moroi et al, or Araoka et al explicitly teach the respective hydrogen and oxygen chambers, pH sensor positioning, or function of the system as claimed.

Kitajima et al is cited as teaching what is conventional in the art of electrolysis, namely that anode and cathode are commonly positioned in respective chambers separated by a membrane (Figures), and that pH sensors are conventionally used in monitoring pH of the electrolyte, and in allowing pH control systems to control pH by adding electrolyte. (Column 3, lines 32-59; also Column 5, line 39 - Column 6, line 6)

It would have been obvious to one having ordinary skill in the art at the time the invention was made to further modify the system of Cifaldi by specifically selecting an electrolyzer with respective anode and cathode chambers, as well as pH monitoring and control systems, as taught by Kitajima et al, because Kitajima shows that such systems are conventional and effective for electrolysis of water. Such selection would only result in the predictable outcome of a working electrolysis system.

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15. Claim 38 is rejected under 35 U.S.C. 103(a) as being unpatentable over Cifaldi and either Moroi et al or Araoka et al as applied to claims 1-6 and 21-26 above, and further in view of Harada et al. (US 5,690,797)

Cifaldi in view of either Moroi et al or Araoka et al is relied upon for the reasons given above.

None among Cifaldi, Moroi et al, or Araoka et al explicitly teach a water level sensor that functions as claimed.

Harada et al is cited as teaching what is conventional in the art of electrolysis, namely that water level sensors are advantageously positioned in the device to control water levels in the device chambers such that the levels remain essentially constant, by triggering the control system to add water to the electrolyzer. (Abstract; Column 5, line 56 - Column 6, line 11)

It would have been obvious to one having ordinary skill in the art at the time the invention was made to further modify the system of Cifaldi by specifically selecting providing the electrolyzer with appropriate water level sensors to control liquid levels in the electrolyzer, as taught by Harada et al, because Harada et al shows that such systems are conventional and effective for control of water levels in electrolytic hydrogen/oxygen generating systems. Such selection would only result in the predictable outcome of a working electrolysis system. Furthermore, any system that electrolyzes water will deplete electrolyte as it operates. As water is removed from the electrolyte, clearly it will eventually need to be replaced. A skilled artisan would have

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recognized the system of Harada as a convenient automatic means of performing this needed function.

16. Claim 39 is rejected under 35 U.S.C. 103(a) as being unpatentable over Cifaldi, either Moroi et al or Araoka et al, and Harada et al as applied to claim 38 above, and further in view of Araoka et al.

Cifaldi in view of either Moroi et al or Araoka et al and Harada et al teaches systems as described above in addressing claim 38.

None among Cifaldi, Moroi et al, or Harada et al explicitly teaches a water source that is a water tank holding water. Harada et al require such a source for water, but is silent concerning what this source is. (Harada et al, Column 10, lines 33-37)

In addition to the teachings cited above, Araoka et al teach that water is provided to an electrolytic cell for hydrogen production from a water tank 3. (English Abstract)

It would have been obvious to one having ordinary skill in the art to modify the water line 162 of Cifaldi by specifically connecting it to a water tank, as taught by Araoka et al, because a water tank is known in the art to be a convenient means for holding water to be provided to an electrolytic cell as needed, as evidenced by Araoka et al. As water is consumed in the cell, it will necessarily need to be replenished, particularly in embodiments where the water generated by the fuel cell is not collected and returned to the electrolytic cell. A water tank is a conventional, convenient means for doing so.

Allowable Subject Matter

17. Claims 7 and 10 are objected to as being dependent upon a rejected base claim, but would be allowable if rewritten in independent form including all of the limitations of the base claim and any intervening claims.

18. Claims 13-20 are allowed.

Response to Arguments

19. Applicant's remarks filed on 30 July 2008 have been fully considered but they are not persuasive.

Applicant cites the reasons for indicating allowable subject matter in the office action of 19 September 2007 in support of Applicant's position that the independent claims are allowable. Claim 7, which included many limitations in addition to those selected for insertion into the independent claims by Applicant was indicated allowable because the prior art did not teach a system *as set forth* in claim 7, with the numerous limitations listed therein. In the interest of not repeating the lengthy claim verbatim, the phrase "as set forth" was used to indicate that the many combined limitations were the basis for allowability. Clearly, the systems cannot be considered allowable due to the bare recitation of "an electrolyte tank" or "and electrolysis chamber", because Cifaldi clearly includes such structure. As the independent claims are not patentable for the reasons given above, the rejections are maintained.

Conclusion

20. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Dr. Jeffrey T. Barton whose telephone number is (571)272-1307. The examiner can normally be reached on M-F 9:00AM - 5:30PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Nam Nguyen can be reached on (571) 272-1342. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Nam X Nguyen/
Supervisory Patent Examiner, Art Unit 1753

JTB
2 January 2009